

Sodium tris(acetylacetonato- κ^2O,O')cobalt(II)Xiaobao Li,^a Ghezai Musie^{a*} and Douglas R. Powell^b^aDepartment of Chemistry, University of Texas at San Antonio, 6900 N Loop 1604 W, San Antonio, TX 78249-0698, USA, and ^bCrystallography Laboratory, The University of Kansas, Lawrence, KS 66045, USA

Correspondence e-mail: gmusie@utsa.edu

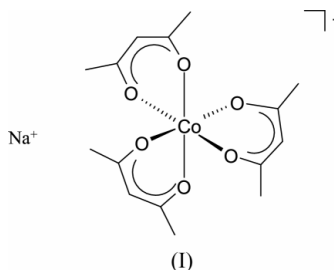
Key indicators

Single-crystal X-ray study
 $T = 100$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.038
 wR factor = 0.119
Data-to-parameter ratio = 14.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $\text{Na}[\text{Co}(\text{acac})_3]$, where acac is acetylacetonate ($\text{C}_5\text{H}_7\text{O}_2$), crystallizes in the rhombohedral space group $R\bar{3}c$. The sodium ion is found to sit on a 32 crystallographic site. The cobalt complex is located on a site of $\bar{3}$ symmetry.

Comment

Metal carboxylate complexes have been the subject of interest for many years (Oldham, 1968; Doedens, 1976). Although a great deal of work has been reported on the synthesis and characterization of various cobalt(III) complexes of the acetylacetonate ligand, only a few cobalt(II) complexes have been reported (Cotton & Elder, 1965, 1966). As part of a general method for the preparation of stable cobalt(II) complexes, we prepared $\text{Na}[\text{Co}(\text{acac})_3]$, (I), and its crystal structure is presented here.



Experimental

A solution of $\text{Co}(\text{acac})_3$ dissolved in methanol was added to a methanol solution of NaBH_4 . After stirring the reaction mixture for 2 h, the solution was concentrated by removing the solvent *in vacuo*. Light-brown needle-shaped crystals were formed by vapor diffusion of diethyl ether.

Crystal data

 $\text{Na}[\text{Co}(\text{C}_5\text{H}_7\text{O}_2)_3]$
 $M_r = 379.24$
Rhombohedral, $R\bar{3}c$
 $a = 16.1201$ (19) Å
 $c = 11.715$ (3) Å
 $V = 2636.3$ (7) Å³
 $Z = 6$
 $D_x = 1.433$ Mg m⁻³Mo $K\alpha$ radiation
Cell parameters from 2493 reflections
 $\theta = 2.5$ – 25.9°
 $\mu = 1.03$ mm⁻¹
 $T = 100$ (2) K
Needle, light brown
 $0.50 \times 0.10 \times 0.07$ mm

Data collection

Bruker APEX CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (SADABS; Sheldrick, 2000)
 $T_{\text{min}} = 0.624$, $T_{\text{max}} = 0.931$
6796 measured reflections548 independent reflections
404 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 26.0^\circ$
 $h = -19 \rightarrow 19$
 $k = -19 \rightarrow 19$
 $l = -14 \rightarrow 14$

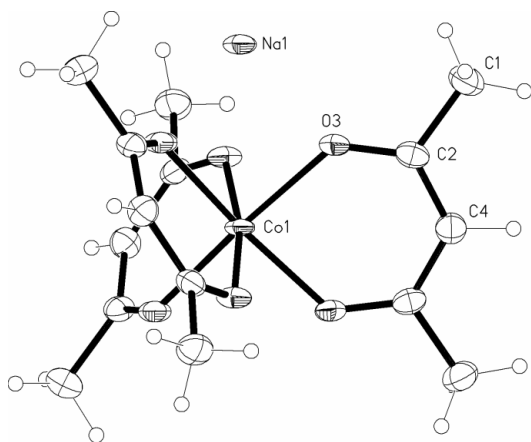


Figure 1
Drawing of $\text{Na}[\text{Co}(\text{acac})_3]$, with 50% probability displacement ellipsoids.

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.038$

$wR(F^2) = 0.119$

$S = 1.25$

548 reflections

37 parameters

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.05P)^2 + 10P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.70 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.33 \text{ e } \text{\AA}^{-3}$$

H-atom positions were refined using a riding model and H-atom displacement parameters were set at 1.2 (1.5 for methyl H atoms) times the U_{eq} value of the bonded atoms.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

The authors are indebted to the Robert A. Welch Foundation for the support of this work under grant AX-1540, and to the National Science Foundation (CHE-0079282) and the University of Kansas for funds to acquire the diffractometer and computers used in this work.

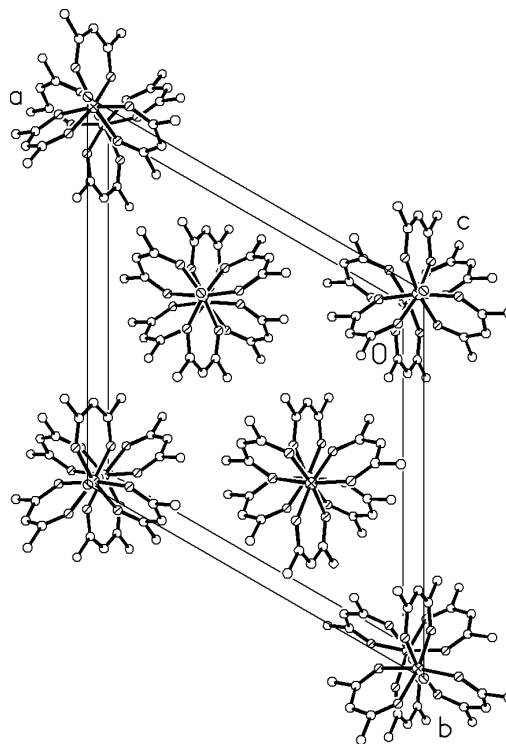


Figure 2
Packing diagram viewed down the c axis. The sodium ion is located at a distance of 2.321 (2) \AA from the nearest O atom of the acac ligand and 2.9287 (6) \AA from the cobalt center.

References

- Bruker (1998). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cotton, F. A. & Elder, R. C. (1965). *Inorg. Chem.* **4**, 1145–1151.
- Cotton, F. A. & Elder, R. C. (1966). *Inorg. Chem.* **5**, 423–429.
- Doedens, R. J. (1976). *Prog. Inorg. Chem.* **21**, 209–231.
- Oldham, C. (1968). *Prog. Inorg. Chem.* **10**, 223–258.
- Sheldrick, G. M. (1997a). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (2000). *SADABS*. University of Göttingen, Germany